

## DOON UNIVERSITY, DEHRADUN

## Mid Semester Examination, 2016 School of Environment & Natural Resources M.Sc. (EVS & NRM), II<sup>nd</sup> Semester

Course: EES - 618: Analytical Techniques & Instrumentations

Time Allowed: 2 Hours

Maximum Marks: 30

Note: Attempt All Questions from Sections A,B,C.

SECTION: A (Short Answer Type Questions/ to be answered in about max 50 words). Attempt any EIGHT questions.

(Marks:  $1.25 \times 8 = 10$ )

- 1. What is the difference between Nephelometry and Turbidimetry?
- 2. Name any two classical qualitative techniques.
- 3. How scattering is related to particle size in a solution?
- 4. What is complexometric titration?
- 5. What is the difference between molar and equivalent conductance?
- 6. What is the principle of Flame Photometry?
- 7. Draw a titration curve between a strong acid and weak base.
- 8. What is potentiometry and what are its applications?
- 9. Define Conductivity?
- 10. List three advantages of modern analytical techniques over classical techniques.

## SECTION: B (Short Answer Type Questions to be answered in about 100 words). Attempt any FOUR questions.

(Marks:  $2.5 \times 4=10$ )

- 1. Differentiate between Reference and Indicator electrode.
- 2. What are the light sources used in UV-Vis spectrophotometers?
- 3. Write a note on Potentiometric titrations.
- 4. Explain Kjeldahl Analysis with reactions involved.
- 5. What are the different parameters which affect photometric measurements?

SECTION: C ( Medium Answer Type Questions to be answered in about 150 words). Attempt any TWO questions.

(Marks:  $5 \times 2 = 10$ )

- 1. Describe principle, instrumentation, and applications of UV Spectrophotometery.
- 2. Explain the principle, instrumentation and applications of potentiometry.
- 3. What is Conductometry? Explain its principle, instrumentation and applications.