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DOON UNIVERSITY, DEHRADUN

Mid Semester Examination, 6th Semester, 2018

Academic Year 2017-18 (Even Semester)

School of Physical Sciences (SoPS) Department name: Chemistry Programme Name: Integrated M.Sc. 5 Years (Chemistry)

Course Code: CYD-351 Course Title: Industrial Chemicals and Environment

Maximum Marks: 30 Time Allowed 2.00 Hours Note: Attempt All Questions from Sections A, B and C. (All terms have their usual meaning) (Marks: 6) SECTION: A 1. Why is He not produced by cryogenic separation of air? What are the sources of He and impurities that needs to be removed in pre-cleaning step of gas? 2. Which method/technique is used in chlor-alkali industry? Explain the technique with chemical reactions involved. [2] 3. Give an account of large scale production of hydrogen. [2] SECTION: B (Marks: 12) 4. What are the applications and hazards of sulphuric acid? What precautions must be taken while handling the chemical? [3] 5. Why is acetylene considered as most versatile fuel? Give its applications and storage conditions. [3] 6. Give a brief account of industrial production, uses and storage of fluorine. [3] 7. What is the commercial name of carbonyl dichloride gas? Discuss its production, uses and storage. [3] SECTION: C (Marks: 12) 8. Write a short note on environmental hazards of: $[1 \times 3]$ (a) Chlorine (b) Carbon monoxide (c) Sulfur dioxide 9. Give an account of large scale production, applications and hazards of nitrogen, oxygen

and argon.

10. Explain the industrial production and applications of:

 $[1 \times 3]$

- (a) Aqua fortis
- (b) Lye
- (c) Hydrochloric acid