

14/12/16



DOON UNIVERSITY, DEHRADUN

End Semester Examination, First (Odd) Semester, 2016–2017

Department of Chemistry, School of Physical Sciences (SoPS)

Integrated M.Sc. 5 Years (Chemistry)

Course: Skill Enhancement Course–I

Course Code: CYS–201

Time Allowed: 03 Hours

Maximum Marks: 30

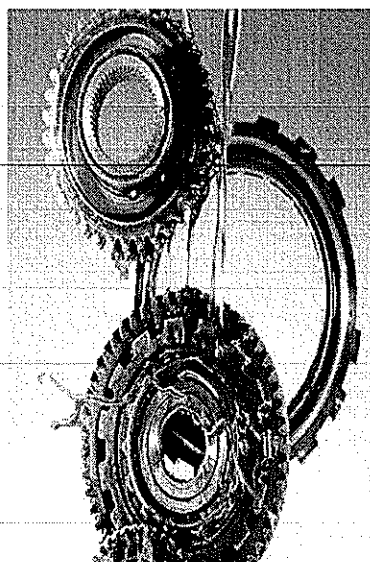
Note: Attempt All Questions from Sections A,B and C.

SECTION: A

(Marks: 6 Q × 1 = 6)

[1] Comment on the following image :

[1]

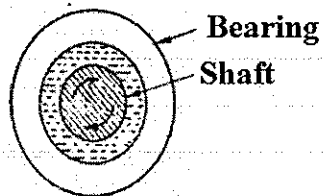


[2] Explain flash point.

[1]

[3] What are the advantages of unleaded petrol? [1]

[4] What kind of lubricant will be suitable for the following? Explain. [1]



[5] Write a short note on Anti-Knocking agents with example. [1]

[6] A gasoline has the same knocking characteristics as a mixture of *iso*-octane and *n*-heptane containing 20% by volume of *n*-heptane. What is the octane number of gasoline. [1]

SECTION: B

Marks: 12

[7] (a) What is Viscosity Index? [1]

(b) Explain the reason why cracking has to be carried out in the absence of air. [1]

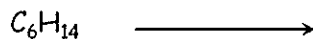
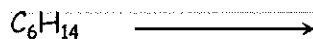
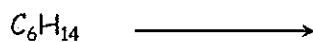
(c) Write a short note on LPG. [1]

[8] Define each of the following terms and give an example of each: [3]

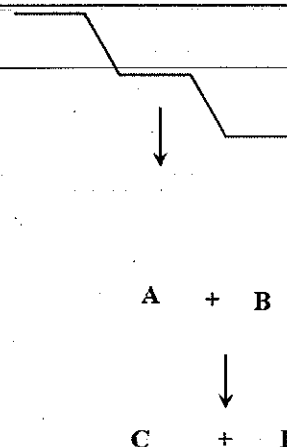
(a) Thermal Cracking (b) Catalytic Cracking (c) Reforming

[9] [3]

(a) What are various possibilities of cracking of C_6H_{14} , a hydrocarbon molecule?



(b) Also explain the mechanism of cracking of the same molecule by filling in the blanks in following.



- [10] A coal sample (weight = m gms) was subjected to ultimate analysis. After that experiment was over, it was noticed that increase in weight of anhydrous calcium chloride tube was z gms and increase in weight of KOH tube was y gms. Write the chemical reactions, which take place during the experiment related to this analysis. Considering this data, fill in the blanks: [3]

44 gm of CO_2 contain = gm of carbon

1 gm of CO_2 will contain = gm of carbon

y gms of CO_2 will contain = gm of carbon

% Carbon = _____

18 gm of H_2O contain = gm of hydrogen

1 gm of H_2O will contain = gm of hydrogen

z gm of H_2O will contain = gm of hydrogen

% Hydrogen = _____

SECTION: C

Marks: 12

- [11] Differentiate between *Proximate Analysis* and *Ultimate Analysis* of Coal. Also explain the reasons why ash-forming constituents in coal are undesirable. In context of *Proximate Analysis*, fill in the blanks: [2+2+2]

% moisture = _____ × -----

% Volatile matter = _____ × -----

% Ash = _____ × -----

% Fixed carbon = _____

- [12] Write short notes on:
 (i) Gross Calorific Value and Net Calorific Value
 (ii) Classification of Lubricants with Examples
 (iii) (a) *Water Gas* (b) *Carbonization*